

# **Challenges on Scaling-Up Heterogeneous Reactions**

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**02 Dec 2024**



# Hovione is a science-based company providing products and services for the Pharmaceutical Industry

- Founded in 1959
- Our customers are **Biotechs, medium, speciality and large pharma**

Drug  
Substance

Drug product

Particle  
engineering



# Manufacturing capacity spread over three continents



- Hovione Sites
- Hovione Offices
- R&D Center



**NEW JERSEY, USA**  
Manufacturing facilities small volume  
R&D Labs, Kilo and pilot plant  
Sales and marketing for North America



**LOURES, PORTUGAL**  
Manufacturing facilities including  
R&D Labs, Kilo and Pilot plants



**CORK, IRELAND**  
Manufacturing facilities

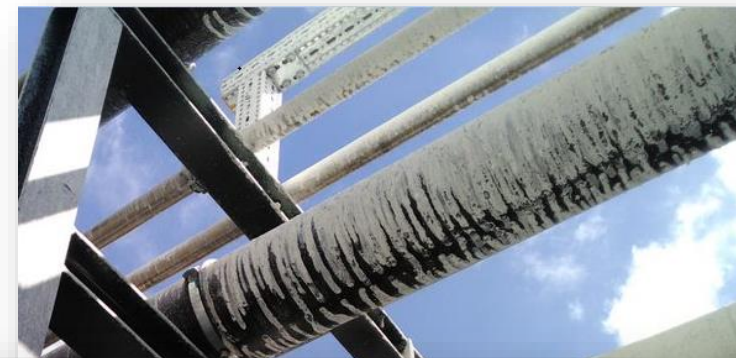


**MACAU, CHINA**  
Manufacturing facilities

# The incident

On 24<sup>th</sup> September 2021

- After reagent addition, a thermal runaway event occurred from 22°C to 46.7°C
- H<sub>2</sub> off-gassing with overpressure of 0.6 bar and foaming
- 80 L of triphasic reaction mixture was ejected through the vent line
- No personal injuries and no property damage
- The affected area was isolated and properly cleaned



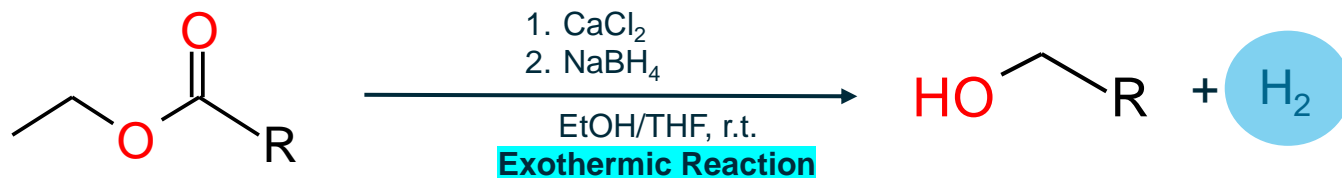
## Process safety incident

Classified as “Potential Critical” (due to severity of the thermal runaway in case of failure of preventive measures in place)

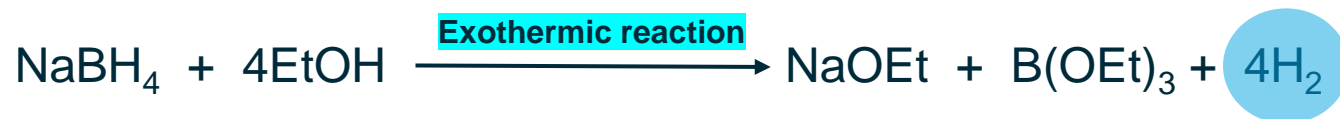


# Chemical Process

## Main reaction - Ester reduction into an alcohol



## Secondary reaction - Reduction of ethanol (ethanolysis)



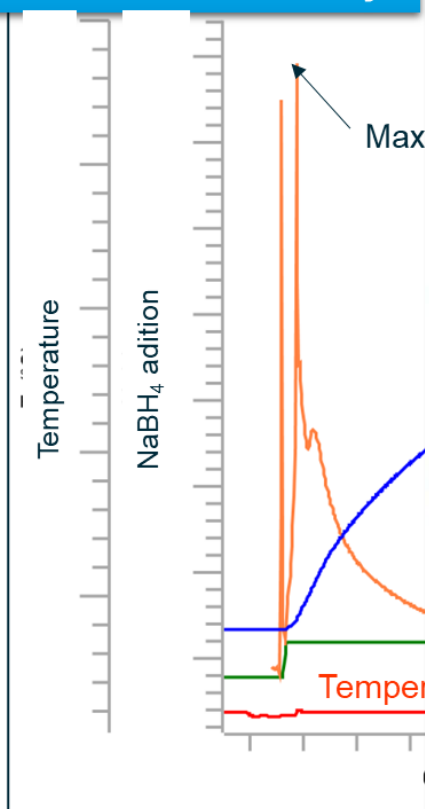
## Important chemical features

- Excess of NaBH<sub>4</sub> to increase process yield
- CaCl<sub>2</sub> act as a catalyst to activate NaBH<sub>4</sub>
- Heterogeneous mixture - very low solubility of NaBH<sub>4</sub> and CaCl<sub>2</sub> in EtOH/THF
- Hydrogen off-gassing results from both main and secondary reactions

# Process Safety studies

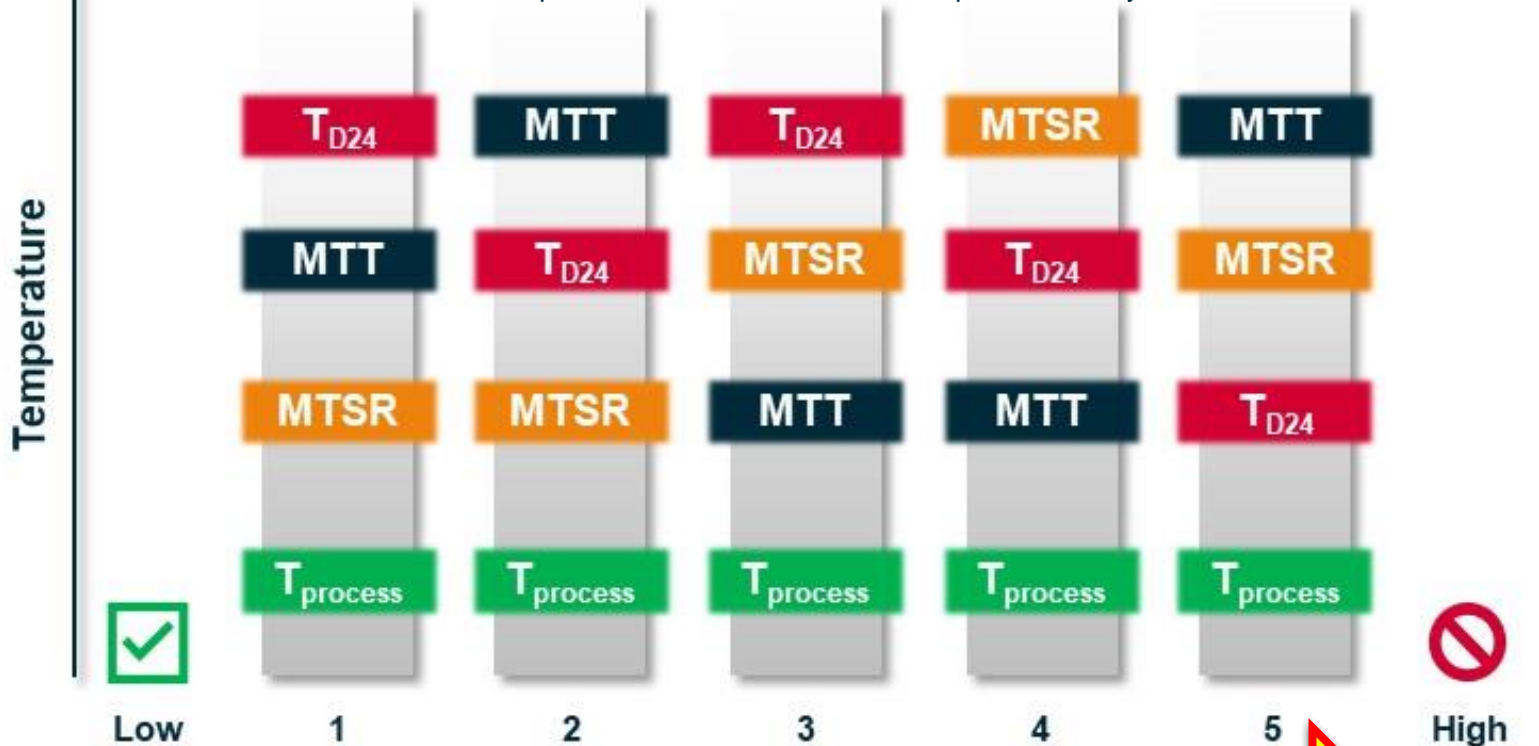
## Thermal Hazard

### Reaction Calorimetry



### CHEMICAL REACTIONS CRITICALITY CLASS

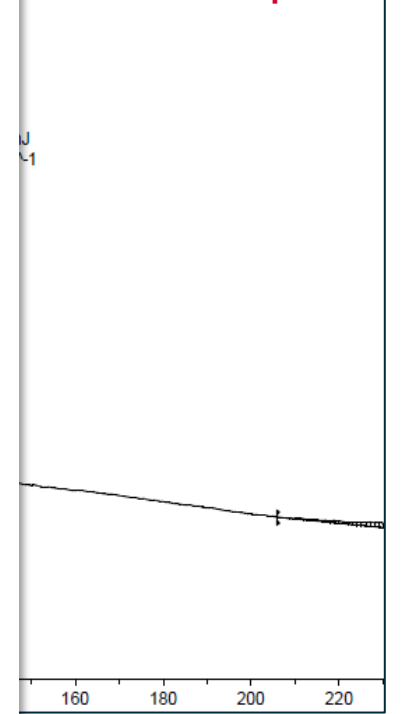
MTT – Maximum technical temperature; MTSR – Maximum Temperature for Synthesis Reaction



**Risk of Thermal Runaway and Overpressure**

### Thermal Stability by DSC

After reaction completed



## Additional studies came out from the HAZOP

How can we decrease the heat accumulation?

Can we increase the  $T_{D24}$ , using a different approach?

What if SRM or  $\text{CaCl}_2$  charge fails?

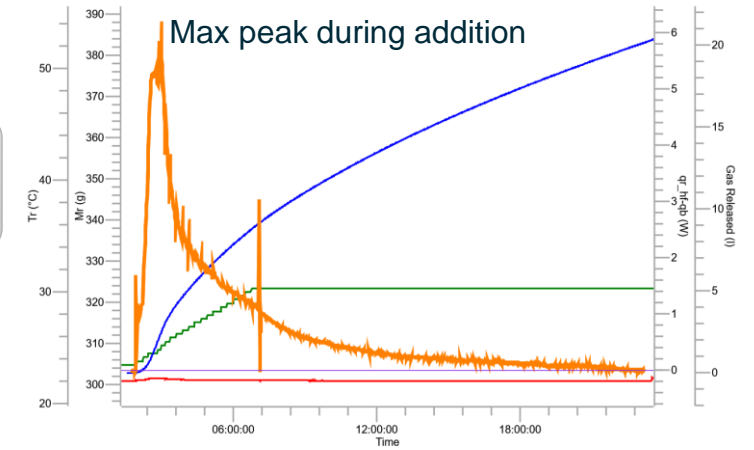


## Additional studies came out from the PHA

How can we decrease the heat accumulation?

Can we increase the  $T_{max}$  using a different approach?  
NaBH<sub>4</sub> addition stepwized in 5h reduced the accumulation from 33% to 20% D24,

What if SRM or CaCl<sub>2</sub> charge fails?





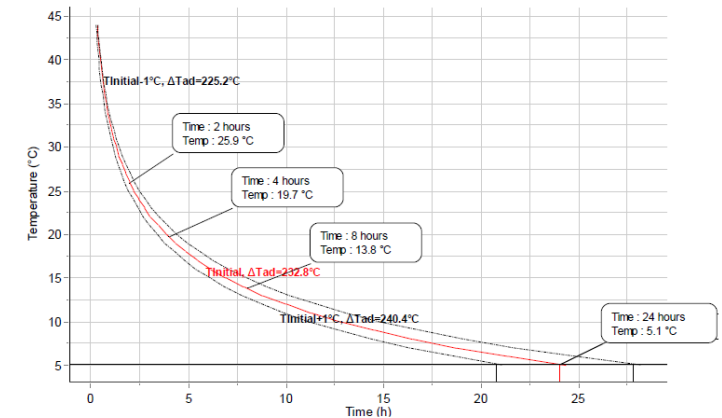
## Additional studies came out from the PHA

How can we decrease the heat accumulation?

Can we increase the  $T_{D24}$ , using a different approach?

- New  $T_{D24}$  (from kinetic studies) is 5°C

What if SRM or  $\text{CaCl}_2$  charge fails? The secondary reaction had more impact on the heat and gas release than the main reaction



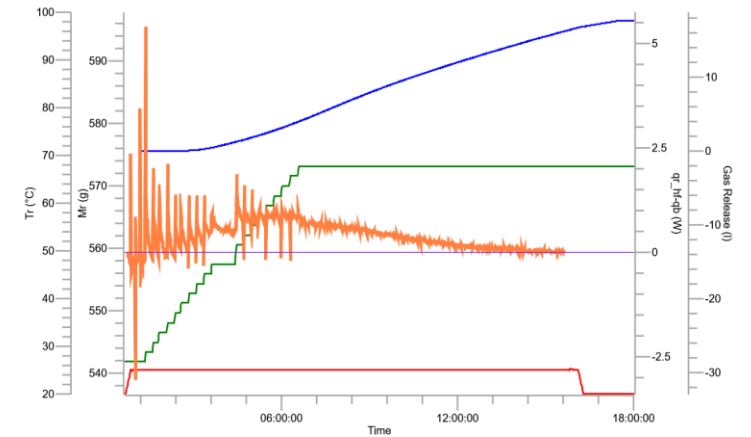
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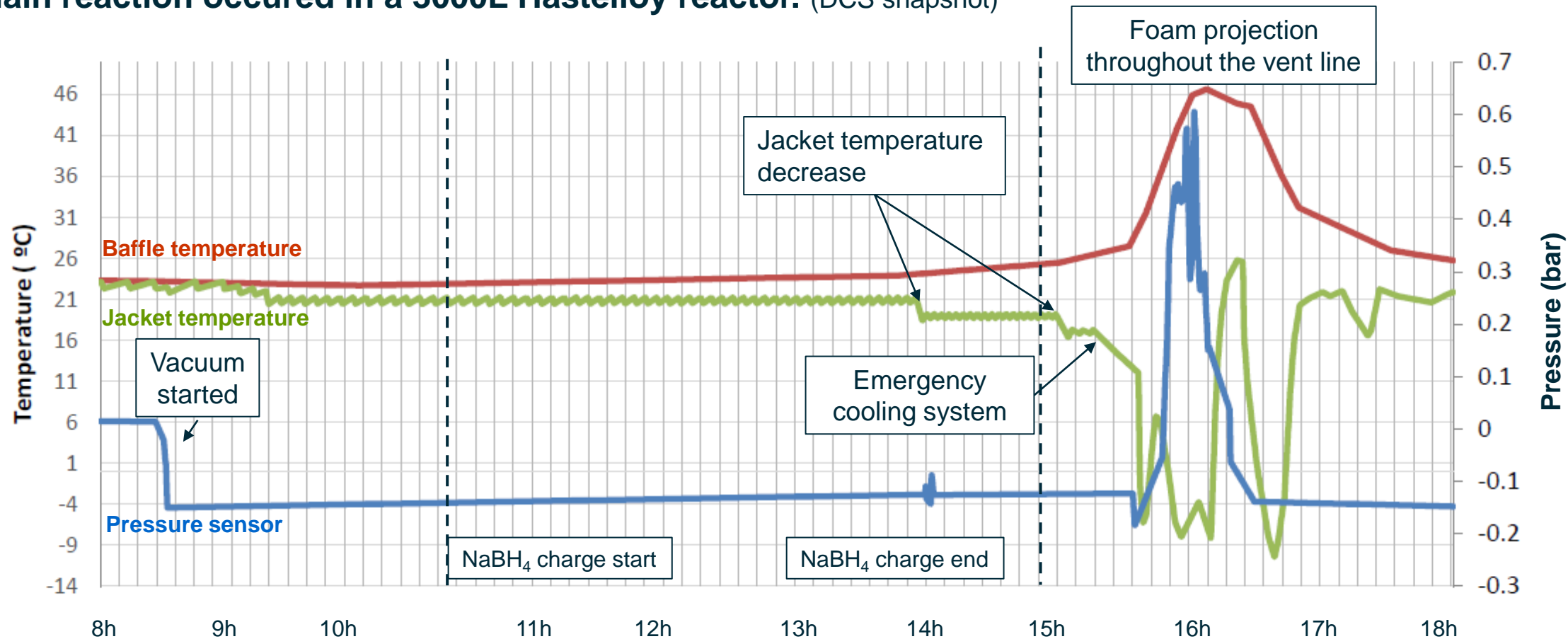
What if SRM and  $\text{CaCl}_2$  charge fails?

Decrease of 74% on the heat release



# The incident

Main reaction occurred in a 3000L Hastelloy reactor. (DCS snapshot)

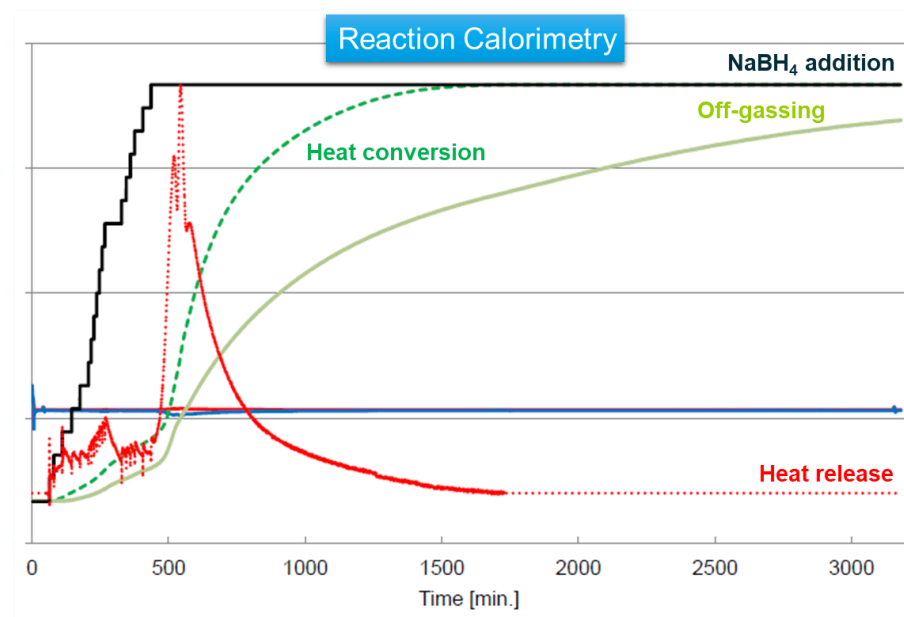


The temperature peak unexpectedly occurred 1h after the NaBH<sub>4</sub> addition

# Incident investigation

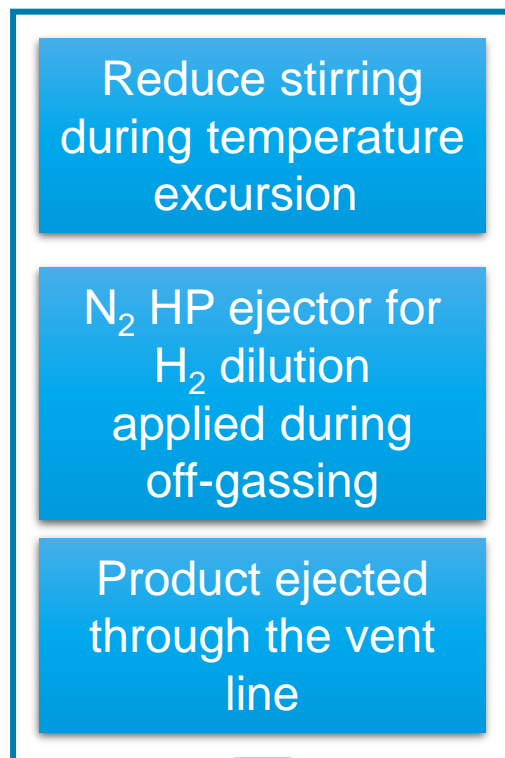
## Heterogeneous nature of the reaction mixture

- Mixing and stirring have impact on the heat and gas release profile with high variability on the process safety data

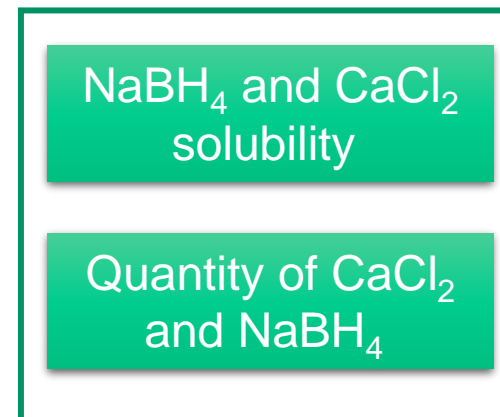


# Incident investigation

## Immediate Causes



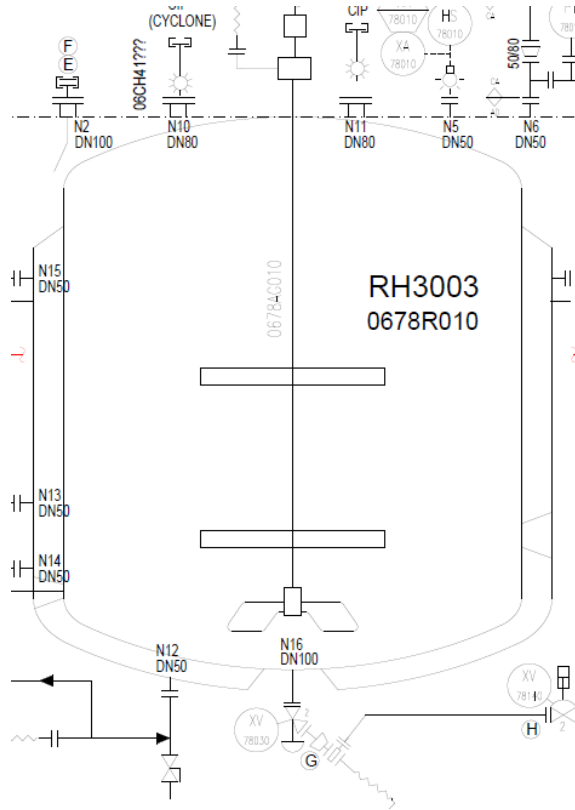
**Operational Causes**



**Process Causes**



# ROOT CAUSE – Training and performance assurance



Decrease of stirring speed during the temperature excursion to avoid foam

Lack of awareness of side effects (heat accumulation) and pressure build up.

Inefficient temperature control by building thermal system. HAZOP identified requirement for double recirculation pump, but manual controlled system.

## ROOT CAUSE – Management of Changes

Heterogenous nature of reaction mixture and lack of full knowledge of the reaction mechanism

Changes in ratio quantities of calcium chloride and sodium borohydride not fully understood.

Rate of sodium borohydride not followed – lack of temperature increase when expected.

## ROOT CAUSE – Process Knowledge Management



H<sub>2</sub> dilution in normal vent done using N<sub>2</sub> ejector. High nitrogene flow produces higher vacuum in Venturi system.

Release system not prepared for foam formation. Emergency vent prepared with Knock out tank but not the normal vent, only design for one phase release.

## Highlights and Strategy after the incident

Workflow defined to evaluate criticality of all reactions and those classified as class 4 and 5 are subjected to HAZOP assessment

Evaluate vacuum effects on reaction with high rate of gas release

Reinforce operators reaction in case of unexpected situations

Heterogeneity of reaction medium must be evaluate in HAZOP assessment

Pre startup safety reviews performed prior to each campaign

# Thank you for your attention



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